₩ McGill

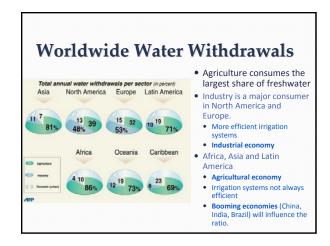
Food for Thought: Issues in Global Water, Food, and Environmental Security

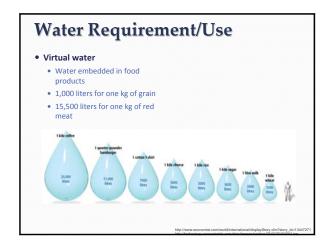
Shiv Om Prasher

James McGill Professor Bioresource Engineering McGill University, Montreal, Canada

Humanity's Top Ten Challenges Energy Water Food Environment Poverty Terrorism & War Disease Education Democracy Population Compiled by Dr. R. E. Smalley - University Professor, Rice University, smalley.rice.edu/Houston Museum.pdf - 1996 Nobel Prize winner in chemistry Kalita (2011)























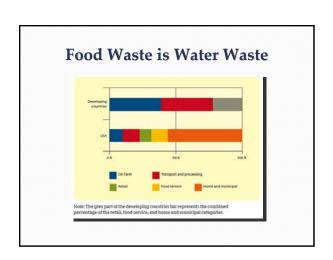


Water and Land Requirements for Vegetarian vs Non-Vegetarian Diets

Vegetarian Non-Vegetarian

- A vegetarian diet can be A meat-based diet requires produced with 900-1,200 m³ well over 2,000 m³ of water of water per person per year per person annually (≈ 100% ♠)
- A vegetarian diet requires no A diet, with 15 percent animal more than 700-800 m² of foodstuff, requires 1,100 m² land per capita /capita (≈ 50% ♠)

A western-type diet, with its high proportion of meat consumption and dairy products, requires up to 4,000 m²/capita (> 500% \spadesuit)



Super Absorbent Polymers or Hydrogel

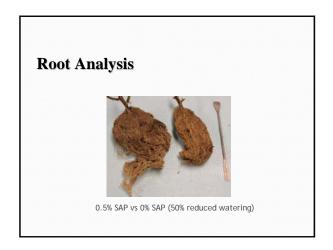
- •Retain Water
- Retain nutrients

How does it work?

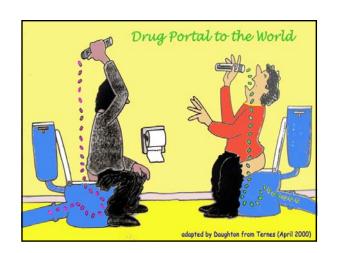
Supply on demand

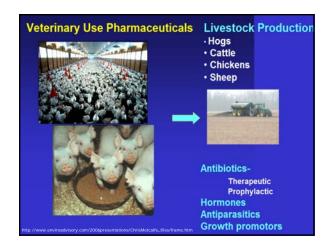




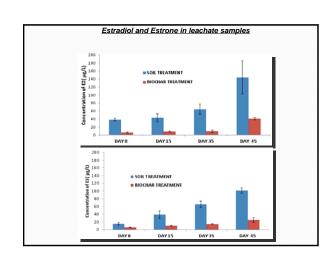


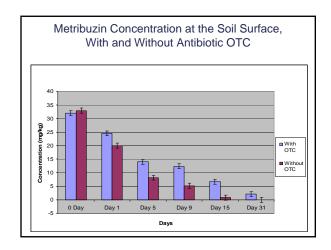








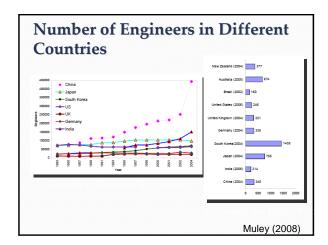


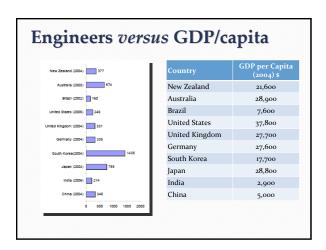


Current Projects

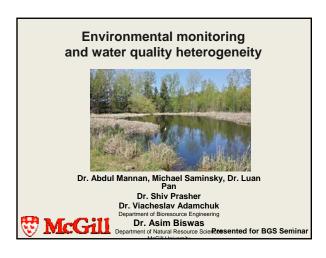
- Biochar and other Additives
 - Effect of biochar on the fate and transport of antibiotics and
 - Pesticide persistence and movement in soil affected by biochar
 - Effect of biochar amendments on the fate and transport of pathogens
 - Use of super absorbent polymers in agriculture agronomic and environmental impacts
- Computer Modeling Field and Watershed scale

 - Antibiotics, Hormones, Pesticides, Fertilize
 SWAT, DRAINMOD, HYDRUS, RZWQM
 - · Stakeholder engagement in physical modeling
- Development of Best Management Practices (BMPs)
- · Water Quality Monitoring of Freshwater Bodies FishNet







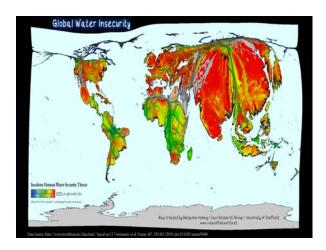


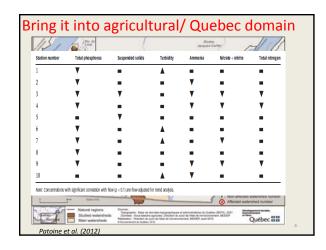
Overview

- 1. Global & local water quality
- Environmental monitoring in small ponds
 Agricultural nutrient loading
 Temporal dynamics

 - c) Spatial Dynamics
- Acoustic sensory telemetry
 a) System components
 b) Sensors

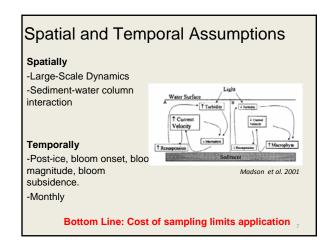
 - c) Automated testing platform
 - d) Tests
- 4. Conclusions

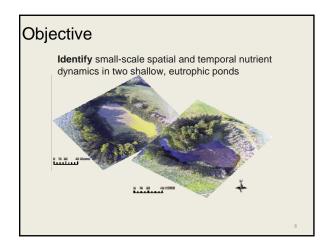


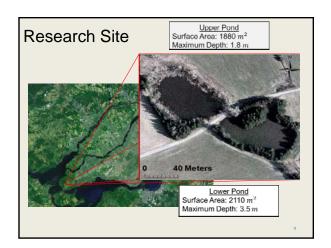


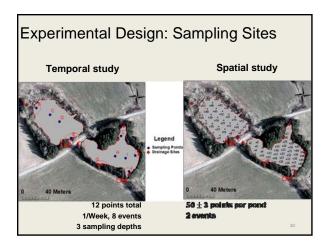
Current state of WQ monitoring					
Method	Pros	Cons			
Manual	•dynamic (can sample multiple spots) • Relatively easy •Well-developed	•Labor-intensive •can't be automated •Expensive •representative?			
permanent/ semi-permanent structures	•automated •real-time •remote	•static •requires maintenance •can be expensive •representative?			
Remote Sensing	low-cost when infrastructure exists actively advancing field	potential for high uncertainty expensive to build, run and operate difficult to interpret			
		5			

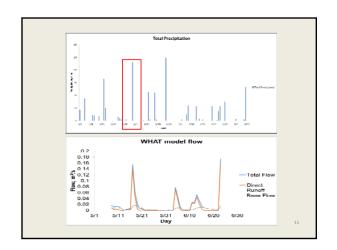
Transition to project

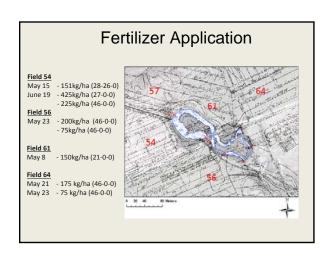


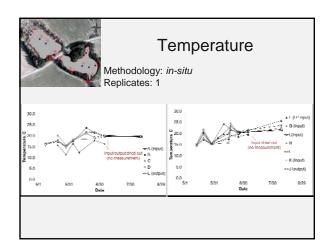


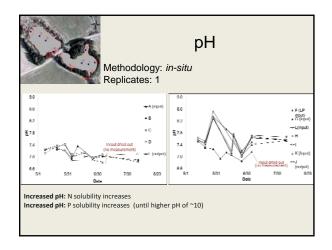


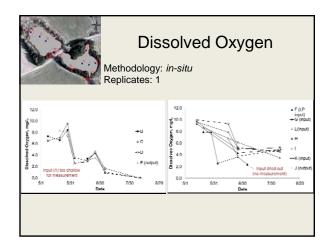


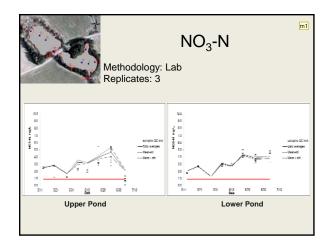


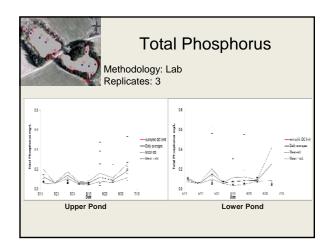


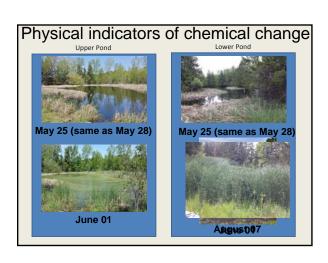






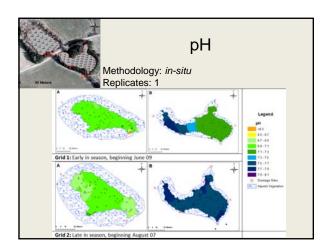


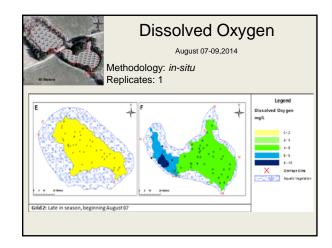


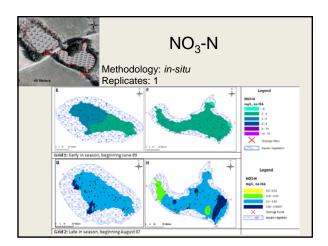


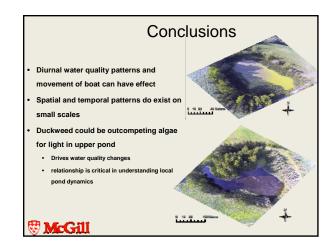
add methodology (in-situ vs lab) and # of reps per point (1 vs 3) m1

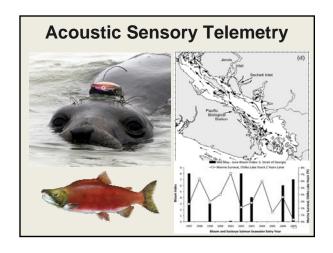
make 3 lowercase Michael, 1/22/2015

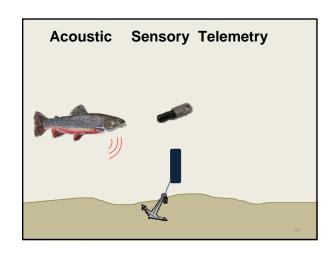


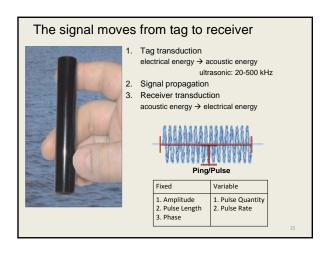


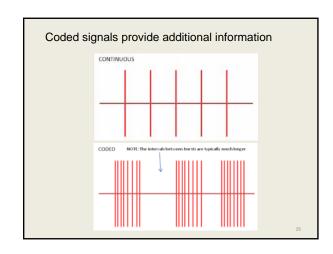


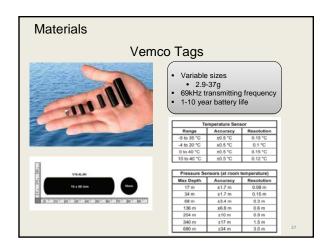






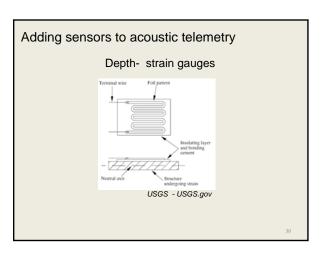


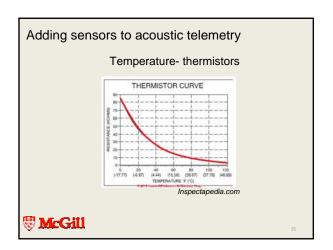


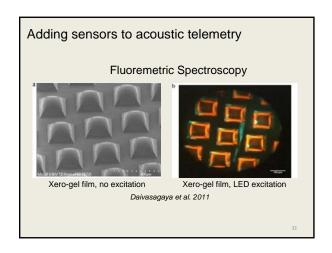


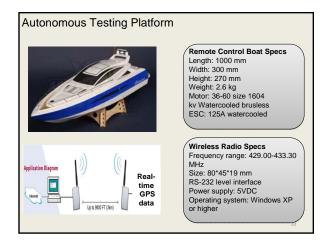


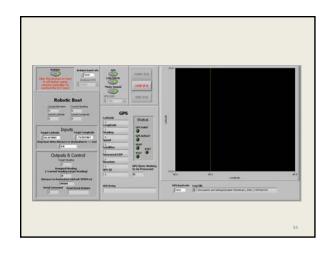
Advantages Disadvantages Continuous sampling Automated Low-Cost Dynamic Non-disruptive Disadvantages Representative? - Fish avoidance

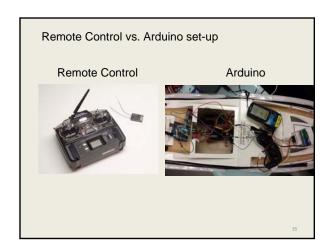


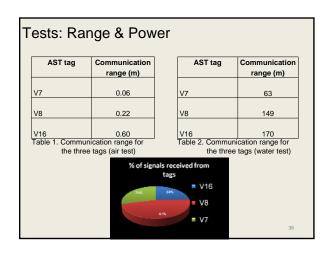


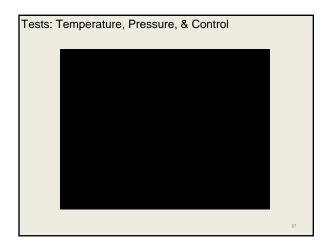












Conclusion

- · Spatial heterogeneity exists even at smallscales
- Improved monitoring technologies needed to effectively assess water quality dynamics

Thank You!

Dr. Viacheslav Adamchuk Dr. Shiv Prasher Dr. Asim Biswas Vemco Systems Natural Sciences and Engineering Research Council

Acoustic Sensory Telemetry

Picture References

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- Nature paper: http://www.nature.com/nature/journal/v467/n7315/pdf/nature09440.pdf
- $Se diment-nutrient\ dynamics\ http://www.jlakes.org/web/Interaction-watermovement-sediment-macrophyte-H2001.pdf$
- http://web.fisheries.org/proofs/tel/Sec7.1_pinco <u>ck.pdf</u>
- http://vemco.com/wpcontent/uploads/2012/11/acoustic telemetry.pd
- http://adamchukpa.mcgill.ca/bgstech/bgstech-p resentation/1402.pdf
- http://adamchukpa.mcgill.ca/bgstech/bgstech_p resentation/1411.pdf
- http://adamchukpa.mcgill.ca/bgstech/bgstech_p resentation/1404.pdf

Fluorescence based Optical pH Sensors

Dr Abdul Monnon, Dr Mark Andrew, Dr Shiv Prosher and Dr Adamciock Department of Chewistry & Department of Bio-vessource Engineering



Outline

- Introduction
- Objective of Study
- Methodology
- Results
- Future Studies

INTRODUCTION

- pH is a frequently measured parameter of immense importance in many application fields, such as environmental monitoring, Chemical Industries, bio-processing and biomedical diagnostics.
- The measurement of pH is usually performed using the glass electrode. however, the electrochemical method suffers from many drawbacks.

Glass Electrode Limitations

- Susceptible to electromagnetic interference
- Difficulty in miniaturisation.
- Limitations of measurement in aqueous suspensions of organic matter.
- High power requirements.
- Bulky design

ADVANTAGES OF OPTICAL PH SENSORS

- Higher sensitivity and selectivity
- Insensitive to electromagnetic interference.
- No need of reference electrode.
- Highly suitable for low power and remote applications.

Objective of Study

 Develop a optical pH sensor having the following Characteristics

Characteristics	Range
pH Range	5-9 (6-8)
Reproducible	Yes
Resolution	0.06-0.08
Device Stability	➤ 1 month
Dye Stability	Stable
Interferences	Ionic independent

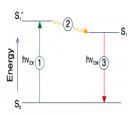
Principal of Optical pH Sensors

- Optical sensors are based on reagents that change their optical properties on with the change in pH of the analyte.
- The most commonly measured optical properties
- Absorption
- · Fluorescence intensity

Fluorescent pH indicators offer better selectivity and sensitivity than absorption-based pH indicators.

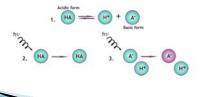
Fluorescence

- Electron excited form ground state by absorption of light
- Fluorescence observed as electron decays – photon release
- Energy lost so light emitted at a longer wavelength



Principal of pH Indicator

PH indicators are typically weak organic acids or bases with distinct optical properties associated with their protonated and deprotonated forms. In most cases, pH sensitivity is based on a fluorescence intensity change with a change in the concentration of hydrogen ions.



Some Fluorescent Dyes and their pH Range

Dyes	pH Range	Excitation	Emission
Fluorescein	5-9	494 nm	521 nm
Resurufin (7-Hydroxyphenoxazin- 3-one)	4-8	460 nm	573 nm
HPTS (pyranine (8- Hydroxypyrene-1,3,6- Trisulfonic Acid, Trisodium Salt))	5-8	450 nm	505 nm
DHPDS (1,3-dihydroxy-pyrene- 6,8-disulfonic acid)	5-8	450 nm	505 nm

Fabrication of Optical pH Sensing Films

- pH sensors consist of a proton-permeable solid matrix in which the pH indicator is encapsulated.
- pH is measured as a function of reversible changes in the fluorescence intensity of the indicator.
- Polymers, Sol-gel materials are widely
- used as immobilisation matrices for pHsensitive fluorescent indicators

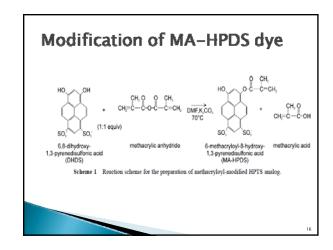
Immobilization of the Flourescent dye on Polymer matrix

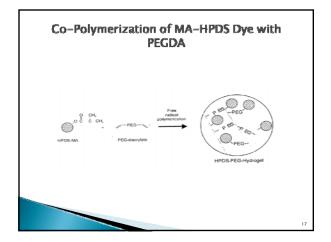
The immobilization of the indicator is probably the most important factor that governs the lifetime and signal stability of the sensor. Poor immobilization causes leaching of the indicator. and, consequently, a drifting of the signal which, in turn, leads to the breakdown of its sensing ability in the extreme case. The most effective immobilization technique is to bind the sensing molecules covalently to the supporting material, which makes leaching virtually impossible under normal conditions

Methodology

Properties and Structure of DHPDS (6,8-dihydroXy-1,3 pyrenedisulfonic acid disodium salt) Dye High quantum yield pH range 5-9 lack of toxicity HO OH Na*

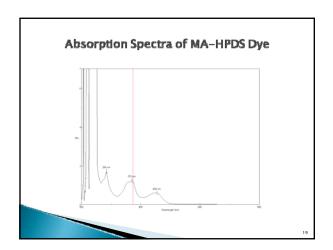
Polymeric Support for Sensor Polyethylene Glycol Diacrylate (PEGDA) H₂C CH₂

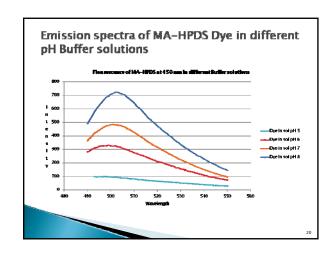


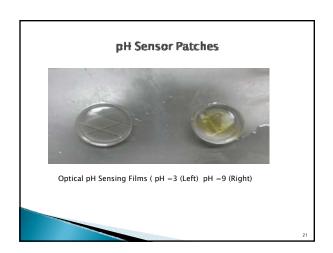


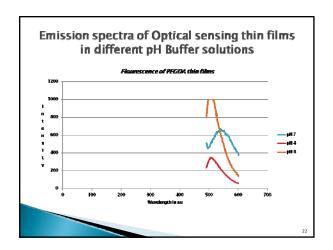
Study of Optical charactaristics of Sensor thin films

- Optical characteristics of the Thin films prepared were analysed on
- UV-Vis Cary 5000 Spectrophotometer
- Flouro-Max-2 Horiba (Capacity to hold solid samples)



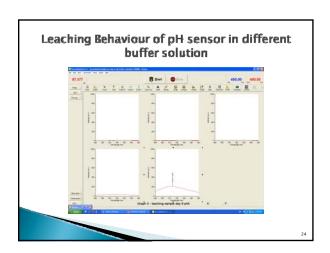






Leaching Behaviour of pH sensor in different buffer solution

Leaching behaviour of PEGDA +Dye Thin films was tested in different buffer solutions for a total period of 9 days. A very small amount of leaching only at pH 8 was observed for all nine days. However, no leaching of dye was observed for buffer solutions of pH 4,5,6 & 7.



Future Studies

- Optimization of thickness of pH sensing thin films for maximum performance

 Development of Optical setup to perform real
- time fluorescence intensity
- Study the behaviour of sensing films in the presence of different ions on the pH measurements.
- Study the effect of temperature on the pH sensing thin films.
- Study the photo-stability and temporal stability of the pH sensing films

THANK YOU FOR YOUR PATIENCE