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Using Ion-Selective Electrodes to Map Soil Properties

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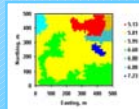
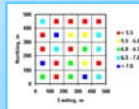
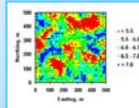
Outline

- Conventional methods of soil mapping
- Automated soil mapping system
- Ion-selective electrodes (pH, K^+ , NO_3^-)
- Laboratory evaluation
- Examples of field mapping
- Agro-economic analysis
- Summary



Conventional Soil Sampling

- Random
- Grid (Systematic) Sampling
 - Grid Point (Cluster) Method
 - Regular (Center)
 - Staggered and Random Start
 - Systematic Unaligned
 - Random
 - Grid Cell Method
- Adaptive
 - By Soil Types
 - By Management Zones



Standard Chemical Soil Test

- Preparation (drying, crushing, sieving)
- Solution
 - 1:1 (Soil Soil pH)
 - 1:2.5 (Nitrate-Nitrogen)
 - 1:10 (Potassium)
- Extraction
 - Water (Soil pH, Nitrate-Nitrogen)
 - SMP Buffer Solution (Buffer pH)
 - Ammonia Acetate Solution (Potassium)
- Measurement
 - Ion-Selective Electrode (pH)
 - Atomic Absorption Spectroscopy (Potassium)
 - Cadmium Reduction Method (Nitrate-Nitrogen)

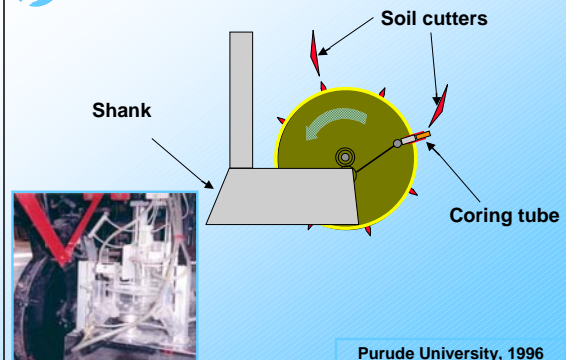


Problem Statement

- The sensing of soil variability is one of the most important steps in site-specific management
- Varying application rates is inappropriate without accurate soil maps
- Obtaining this descriptive information about a field is expensive using conventional methods
- There is a need to develop equipment for mapping chemical soil attributes on-the-go
- Offered technology must be reliable, rapid, simple, inexpensive, repeatable



Soil Nutrients Mapping



Soil Nutrients Mapping

Purdue University, 1996

Direct Soil Measurement

- Preparation
 - Field conditions
- Solution
 - Naturally moist soil
- Extraction
 - Available ion activity
- Measurement
 - Ion-selective electrode

Automated Soil pH Mapping Systems

US Patent No. 6,356,830
Purdue University, 2000

Automated Soil pH Mapping Systems

Veris Technologies, 2002

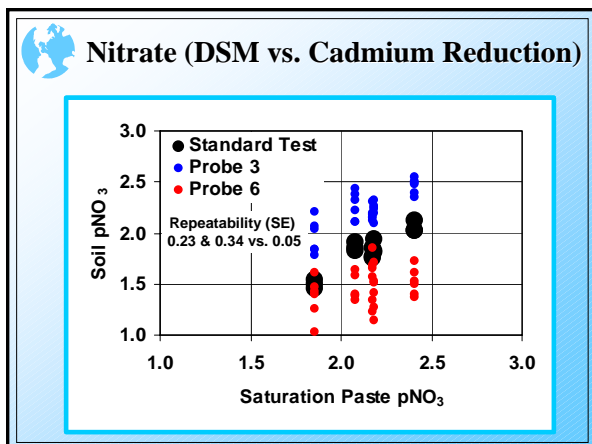
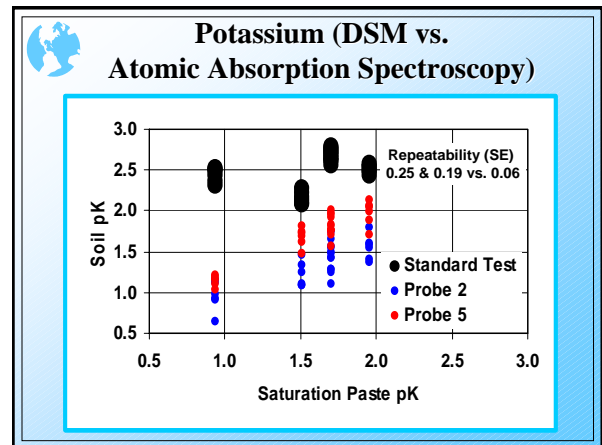
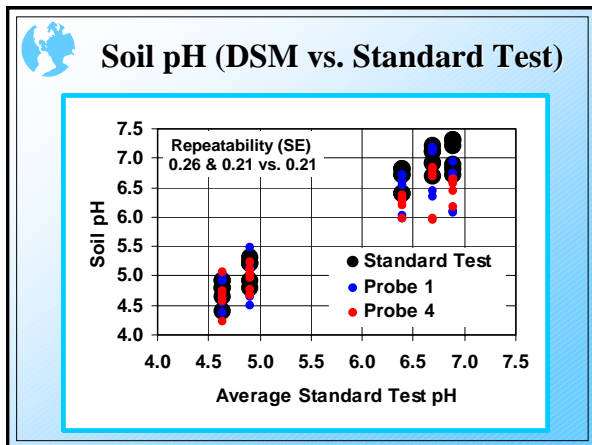
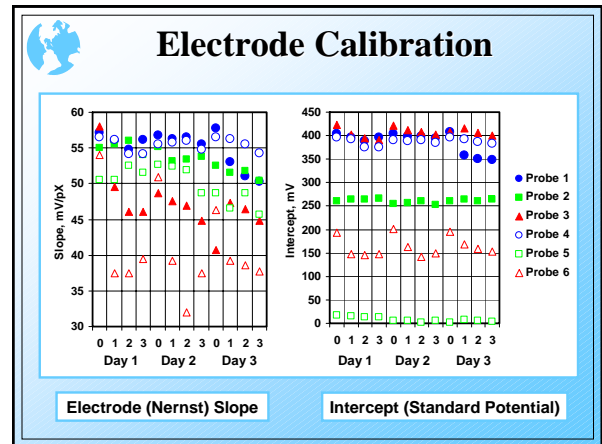
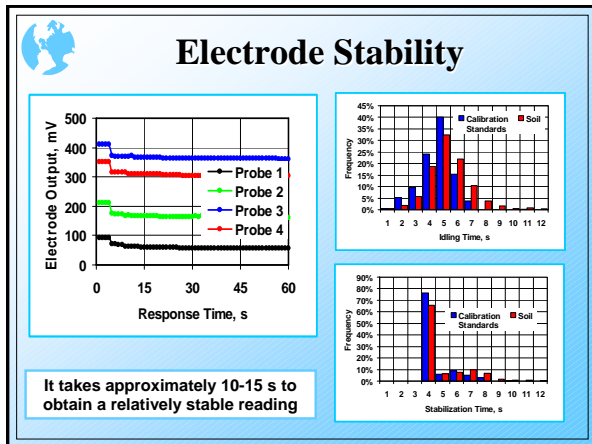
Ion-selective Electrodes

- Combination
- Flat Surface
- Jell-Filled
- Epoxy Body
- Glass (pH)
- PVC Membrane (K^+ , NO_3^-)

Laboratory Evaluation of Direct Soil Mapping

- Stability
- Calibration
- Repeatability
- Accuracy
- Reliability

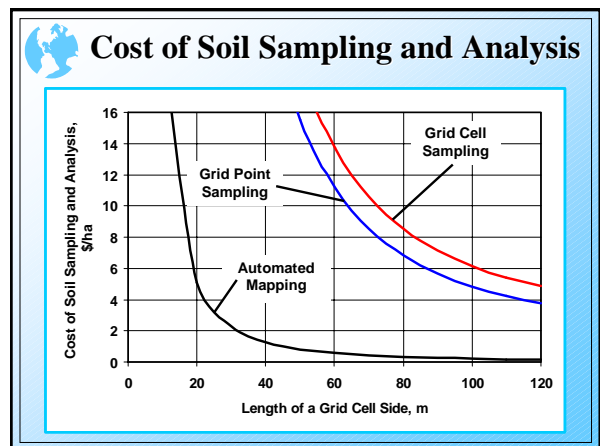
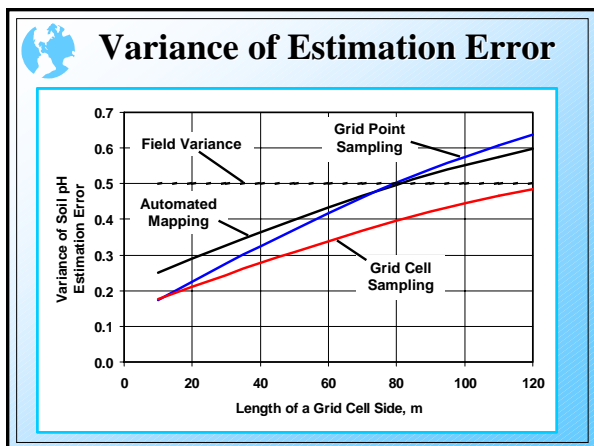
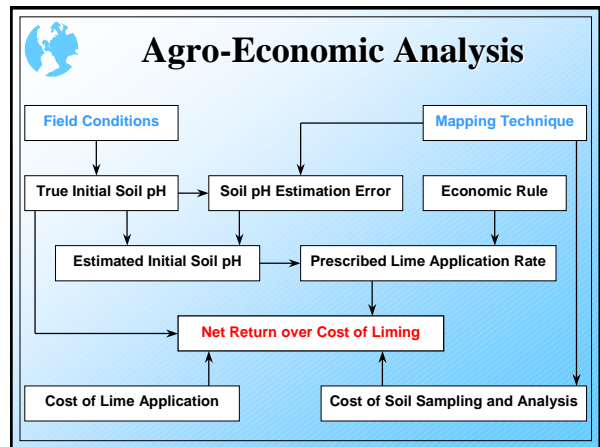
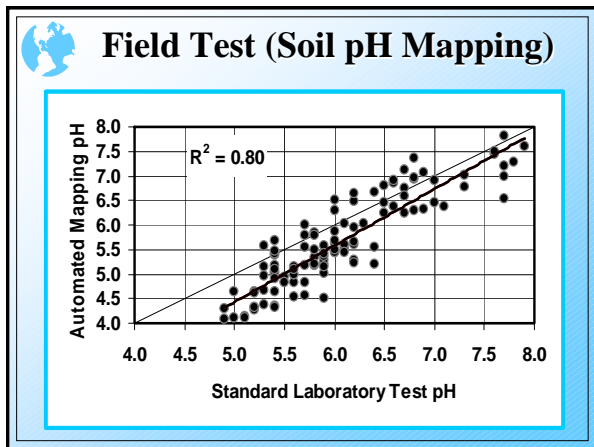
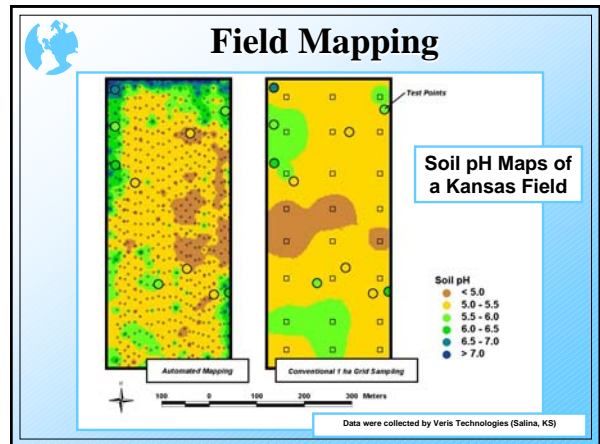
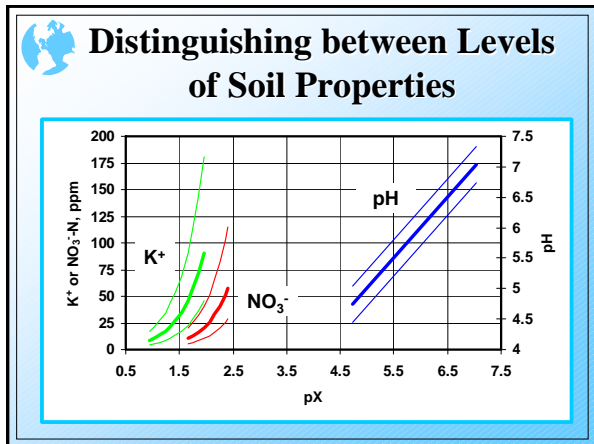
- Crete Silt Loam
- Wymore Silty Clay Loam
- Brocksburg Sandy Loam
- Hall Silt Loam
- Thurman-Anselmo Fine Sandy Loam

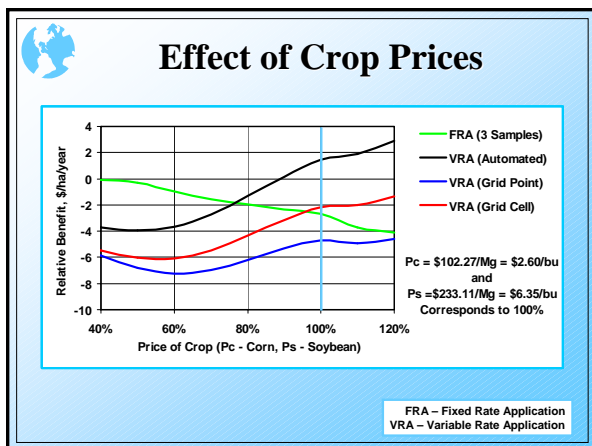
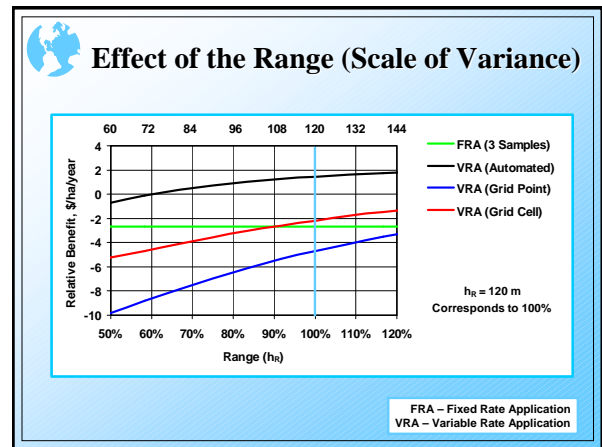
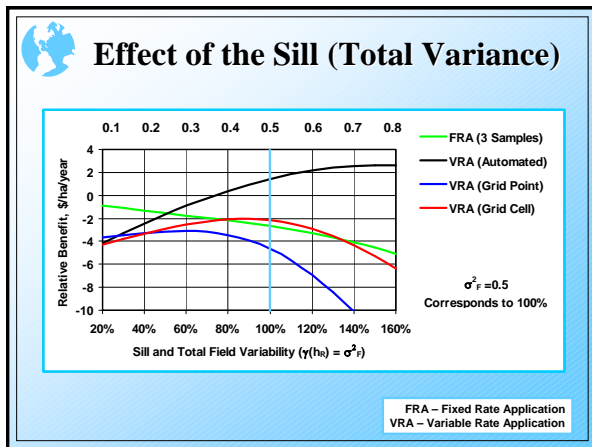
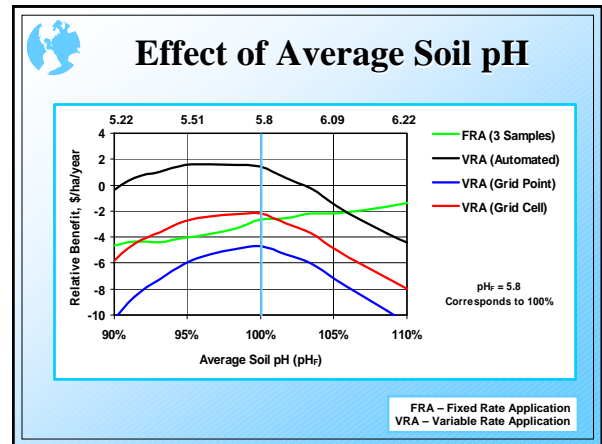
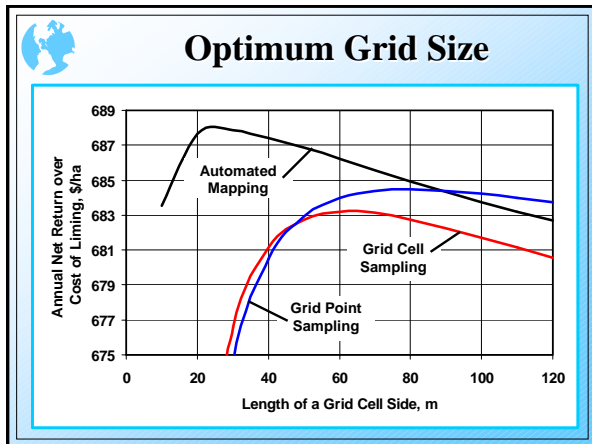


Electrode Accuracy

| | Probe 1 | Probe 2 | Probe 3 | Probe 4 | Probe 5 | Probe 6 |
|---------------------------|---------|---------|------------------|---------|---------|------------------|
| | pH | pK | pNO ₃ | pH | pK | pNO ₃ |
| | Brand 1 | | | Brand 2 | | |
| Individual R ² | 0.89 | 0.40 | 0.30 | 0.93 | 0.67 | 0.03 |
| Daily Mean R ² | 0.96 | 0.61 | 0.60 | 0.97 | 0.83 | 0.08 |
| Mean R ² | 0.98 | 0.89 | 0.89 | 0.99 | 0.95 | 0.56 |

Reference:
 Average Standard Test for pH
 Saturation Paste and Atomic Absorption Spectroscopy for pK
 Saturation Paste and Cadmium Reduction for pNO₃





- ### Summary
- Direct soil measurement of H⁺, K⁺ and NO₃⁻ have root mean squared errors less than 0.3 pX (pH, pK, pNO₃)
 - Automated soil pH mapping is a promising alternative to conventional sampling and analysis methods due to increased sampling density
 - Estimating exchangeable potassium content or buffer pH requires additional information about soil tested
 - Direct automated mapping of phosphorus content remains non-feasible



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