

Mark Morgan Daniel Ess

Agricultural and Biological Engineering Purdue University

July, 1998



Negative Influence of Low pH Acidic ions can be toxic for some plants Limited amount of basic ions (Ca²⁺ and Mg²⁺) Reduced availability of nutrients (N-P-K) Decreased effectiveness of some herbicides Low activity of bio-organisms

D.B. Mengel, 1997

Laboratory Soil pH Measurement

- 1. Calibrate the pH meter over the appropriate range.
- 2. Measure a soil sample into a cup.
- 3. Add pure water to the sample to bring the solution to a weight-to-weight ratio of 1:1.
- 4. Stir vigorously for 5 seconds and let stand for 10 minutes.
- 5. Place electrodes in the slurry, swirl carefully, and *read the pH*.

M.E. Watson and J.R. Brown, 1997



































Estimated Sampling Density (samples / ha)						
	Vehicle Speed,	Distance between	Distance between Passes, m			
	km/h	Samples, m	5	10	20	50
	1	2.2	900	450	225	90
	2	4.4	450	225	113	45
	3	6.7	300	150	75	30
	5	11.1	180	90	45	18
/	8	17.8	113	56	28	11



